Motion of an Adhesive Gel in a Swelling Gradient: A Mechanism for Cell Locomotion

Jean-François Joanny, Frank Jülicher, 1,2 and Jacques Prost 1

¹Physicochimie Curie (CNRS-UMR168), Institut Curie Section Recherche, 26 rue d'Ulm, 75248 Paris Cedex 05, France ²Max-Planck Institut für Physik komplexer Systeme, Nöthnitzerstrasse 38, 01187 Dresden, Germany (Received 2 December 2002; published 25 April 2003)

Motivated by the motion of nematode sperm cells, we present a model for the motion of an adhesive gel on a solid substrate. The gel polymerizes at the leading edge and depolymerizes at the rear. The motion results from a competition between a self-generated swelling gradient and the adhesion on the substrate. The resulting stress provokes the rupture of the adhesion points and allows for the motion. The model predicts an unusual force-velocity relation which depends in significant ways on the point of application of the force.

DOI: 10.1103/PhysRevLett.90.168102

Many cells are able to crawl on a solid substrate to which they adhere. This motion involves, in general, three processes: the formation and protrusion of a thin lamellipod in front of the cell, the adhesion of the lamellipod to the substrate, and its retraction at the rear, pulling the cell forward. The formation of the lamellipod involves the polymerization of cytoskeletal filaments and their cross-linking near the membrane. Motion generation requires a symmetry breaking often associated with the treadmilling of the cell cytoskeleton (the continuous asymmetric polymerization at the leading extremity of the gel and depolymerization at the rear). Typical examples for crawling cells are the motion of fibroblast cells or fish epidermal keratocytes that have been studied in detail. In many cells the cytoskeleton is an actin network and the motion generation involves a complex interaction between actin, myosin II motors, and many other proteins [1-3].

From a point of view of physics, cell motion is closely related to the motion of an elastic gel. Spontaneous motion of gels has been studied in several contexts. A mechanism has been proposed where a gel rolls on a substrate without slipping; the gel is not treadmilling in this case [4]. In order to explain the motion of the bacterium Listeria which is propelled by formation of an actin gel comet, a description has been proposed which is based on the interplay of gel elasticity, polymerization on the surface of the bacteria, and depolymerization at the outer surface of the gel [5]. The coupling between polymerization and motion is due to the stress developed between the actin gel and the bacterium. Biomimetic experiments on the motion of small colloidal spheres induced by an actin comet have been performed to test these ideas [6]. More recently, Bottino et al. [7] have described the crawling of nematode sperm cells on a surface. The cytoskeleton is represented by a gel which is treadmilling and adhering to the substrate. The polarity of the cell leads to a pH gradient that subsequently induces a gradient in internal tension and adhesion to the substrate which are introduced phenomenologically.

Ascaris nematode sperm cells provide an example of motion of a treadmilling gel in a swelling gradient. The motility of these cells has been thoroughly studied experimentally [8,9]. The cytoskeleton mostly consists of an apolar protein, major sperm protein (MSP), that similarly to actin polymerizes to form filaments which can be cross-linked to form a gel [10]. The moving cell forms a thin lamellipod on the substrate and the cell body is dragged behind the lamellipod. There is convincing evidence that polymerization of the MSP network proceeds at the advancing edge of a motile cell [11,12]. Depolymerization occurs in the vicinity of the cell body. ATP is hydrolyzed in the polymerization process but molecular motors apparently do not play a role for motility. The local pH strongly influences the polymerization and depolymerization of the MSP network [13]: an increase of pH enhances the polymerization and a decrease of pH provokes depolymerization of the network. It has been proposed that an influx of protons in the cell body creates a pH source which leads to a pH gradient and thus an asymmetry between the front and the rear of the cell [14].

PACS numbers: 87.17.Jj, 82.35.Gh, 82.70.Gg

The aim of the present Letter is to describe on very general physical grounds the motion of an isotropic elastic gel adhering on a surface by considering the interplay between a self-generated concentration field (pH) that creates a gradient in the gel properties and the adhesion. In order to illustrate the main principles of motion of an adhesive gel in a swelling gradient, we use simplified assumptions. We assume that the main effect of the pH is to change the local equilibrium swelling of the gel and that the advancing gel is more contracted at the rear (our mechanism is, however, more general and does not depend on the sign of the gradient); we furthermore assume that the gel adheres to the substrate by localized adhesion points and that the adhesion is strongly irreversible. This implies that the adhesion points are broken if they are subject to a constraint that exceeds a threshold. Using general elasticity theory, we show that the combination of adhesion and an equilibrium swelling gradient generates stresses in the gel that become large at the rear part and provoke a rupture of the adhesion points. For the sake of simplicity, we use a two-dimensional model that considers only one dimension in the direction of motion and the direction transverse to the plane of motion.

Consider a thin treadmilling gel in a concentration gradient (pH). The situation sketched in Fig. 1 involves a gel of small thickness h that adheres on the substrate over a length $L \gg h$. Behind the adhering part, there is a nonadhering part of the gel of length L'. The pH gradient is stationary in the reference frame of the moving gel. The gel forms at the leading edge by polymerization and it depolymerizes at the rear. In the following, we use the state of the gel formed by polymerization as the reference state for the quantification of elastic deformations: in this state, two points connected by a vector $\mathbf{d}\mathbf{x}$ are at a distance $ds_0^2 = \delta_{ij} dx_i dx_j$. As the polymerization proceeds, each point of the gel is transported by the treadmilling process inside the moving gel and displaced by a vector **u** with respect to the substrate. The gel undergoes elastic deformations resulting in changes in the distance between the two points, which becomes $ds^2 =$ $g_{ij}dx_idx_j$, where $g_{ij}=\delta_{ij}+2\epsilon_{ij}$ is the metric tensor and $\epsilon_{ij}=\frac{1}{2}[(\partial u_i/\partial x_j)+(\partial u_j/\partial x_i)]$ is the displacement gradient (which should not be confused with the strain tensor defined below) [15]. In the reference frame of the gel, a given volume element is transported toward low pH regions as time goes on; its equilibrium swelling changes and it tends to contract. We assume here for simplicity that the equilibrium state of the gel at a given position (a given pH) is isotropic. Swelling of the gel implies that the distance between two neighboring points in the equilibrium state is contracted by a factor Λ that depends only on the local pH, as compared to the reference state in which it was formed. The contraction factor Λ increases from a value $\Lambda < 1$ at the rear of the gel to $\Lambda = 1$ at the leading edge. The equilibrium state of the gel is the state that a small gel element would reach if it were cut from the rest of the gel while remaining in its environment. The metric tensor in the local equilibrium state is $g_{ij}^0 = \Lambda^2 \delta_{ij}$. The strain tensor is defined in a standard way as $u_{ij} =$ $\frac{1}{2}(g_{ij}-g_{ij}^0)=[(1-\Lambda^2)/2]\delta_{ij}+\epsilon_{ij}$. The stress tensor in the gel σ_{ii} can be calculated by assuming again that the gel is isotropic and by introducing the two Lamé coefficients λ and μ such that $\sigma_{ij} = \lambda u_{kk} \delta_{ij} + 2\mu u_{ij}$ [16]. The two Lamé coefficients may themselves depend on the

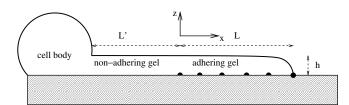


FIG. 1. Schematic representation of the advancing cell on a substrate. The lamellipod is filled by a gel that adheres in the front region of length L. The length of the nonadhering part in the back is L'. The thickness of the lamellipod is denoted h.

local pH but, within a linear elasticity description, here we ignore this variation. Our model for the motion is one dimensional, x is the coordinate parallel to the substrate, and z is perpendicular. The stress tensor can then be written as $\sigma_{ij} = (\lambda + \mu)(1 - \Lambda^2)\delta_{ij} + \lambda \epsilon_{kk}\delta_{ij} + 2\mu \epsilon_{ij}$. The second and third terms in this expression are the standard contributions from linear elasticity theory; the first term, related to the change in equilibrium swelling, plays the role of the tensile stress introduced phenomenologically by Bottino et al. [7].

Locally, the gel is at mechanical equilibrium and the divergence of the stress tensor vanishes $(\partial/\partial x_j)\sigma_{ij}=0$. However, the gel does not relax to its equilibrium swelling since it adheres on the surface. If we assume that the adhesion occurs at the leading edge upon the formation of the gel and that the adhesion points are frozen, the displacement \mathbf{u} vanishes at each point on the surface. The upper surface of the gel is free and the two components of the stress vanish there $\sigma_{iz}(h)=0$.

The elastic deformation is calculated by solving the force balance equation. Inside the gel, we use a lubrication approximation [17]: the variation of the displacement occurs over a distance h in the perpendicular direction (z) much smaller than the distance L in the direction parallel to the substrate, and the derivatives with respect to z are thus much larger than the derivatives with respect to x. We find [18]

$$u_{x} = \frac{\lambda + \mu}{\lambda + 2\mu} \frac{d(\Lambda^{2}(x) - 1)}{dx} \left(\frac{z^{2}}{2} - 2hz\right),$$

$$u_{z} = \frac{\lambda + \mu}{\lambda + 2\mu} (\Lambda^{2}(x) - 1)z.$$
(1)

The deformation in the x direction at the free surface (z=h) is negative towards the rear where the gel contraction is stronger. The "lubrication approximation" is valid everywhere in the gel except in a region of size h both at the leading edge and at the rear of the gel. The distance to the edge of the gel is smaller than h in these regions and the full elasticity equations must be solved. At the rear, the gel tends to contract and the deformation in the x direction becomes positive. The tangential stress on the substrate (z=0) at position x reads

$$\sigma_{xz} = -2\mu \frac{\lambda + \mu}{\lambda + 2\mu} h \frac{d(\Lambda^2 - 1)}{dx}.$$
 (2)

The stress on the surface is negative as the gel is pulled towards the rear part. The tangential stress on the surface is balanced by the force per unit area exerted by the adhesion points. If there is a uniform density ρ_s of adhesion points, the (positive) force exerted by each point on the gel is $f = -\sigma_{xz}/\rho_s$.

So far, we have considered a flat piece of gel of length L adhering on the substrate. At the rear of this piece over a region with a size of the order of the thickness h, the signs of both the deformation and the tangential stress change. A precise calculation of the tangential stress in this region

168102-2

is difficult as it requires a complete solution of the elasticity equations. However, one can get an estimate of the stress using the macroscopic force balance on the adhering gel along the substrate. The total force acting on the gel vanishes. Since there are no external forces acting at the upper surface and at the leading edge, the force balance leads to $\int_0^L \sigma_{xz} dx = 0$. If we neglect the variation of the tangential stress in the region of size h where it is positive, we obtain

$$\sigma_{xz}(x=0) \sim \mu \frac{\lambda + \mu}{\lambda + 2\mu} (1 - \Lambda^2).$$
 (3)

The lubrication approximation is consistent if $h\{[d(\Lambda^2 - 1)]/dx\} \ll 1$ and the stress pulling the gel forward at the back of the gel is much larger than the stress pulling it backwards in the front parts. The tension on the adhesion points is thus largest at the back of the adhering part of the gel.

The adhesion points on the substrate have a finite strength; they rupture if the force exerted by the gel is too large. We consider here that adhesion is irreversible and that there is a threshold for the rupture of an adhesion point (that can be associated to several molecular bonds). The rupture of each adhesion point can be viewed as the escape of a particle from an attractive potential well under the action of an external force (the gel stress). This problem has been studied in detail by several authors [19,20]; here we use the simplified approach that ignores thermal fluctuations which amounts to characterizing each adhesion point by the critical force f_c that it can sustain (this force could also, in principle, depend on pH and on the time scale of the adhesion breakage [20]). The stress exerted on the substrate by the gel being largest at the back of the adhering gel, the adhesion points first break at the back of the gel. The critical adhesion force, using Eq. (3), defines a critical value of the tangential stress and thus a critical value of the swelling ratio Λ_c at which adhesion bonds rupture. If we assume steady state motion, the pH profile and thus also the profile of swelling $\Lambda(x)$ are stationary in the reference frame of the gel. The critical adhesion force then imposes the distance L' between the rear of the gel and the point where the adhesion points rupture when $\Lambda(0) = \Lambda_c$. An important point is that our model does not require any adhesion gradient; the adhesion rupture occurs at the point of strongest tension.

We therefore obtain the picture sketched in Fig. 1 of a gel adhering to the substrate in the front part over a length L but not adhering in the rear over a length L'. The adhering gel is under stress while the nonadhering gel can relax to its local equilibrium swelling characterized by $\Lambda(x)$. The picture is not changed qualitatively if the critical force of the adhesion points f_c depends on pH; only the precise value of L' changes. The concept of rupture of discrete adhesion points implies that the motion is not steady but occurs in jumps. As the depolymerization (and the polymerization) proceeds, the force on

the most strongly loaded adhesion point increases further until the critical value is reached, which provokes rupture. When an adhesion point ruptures, the gel relaxes rapidly and moves in the direction of motion. We implicitly assume here that the rupture is irreversible: adhesion points do not reform immediately and the kinetics of formation of the adhesion points is slow compared to the depolymerization kinetics.

The model is consistent if the internal stress is large enough to induce rupture of the adhesion points (L' must be positive) and if the adhesion points can form (L must be larger than h); otherwise, the gel gets stuck on the substrate. Our simplified description relies on the use of a two-dimensional model. For a real three-dimensional gel, collective effects such as detachment waves similar to those observed in friction problems could be expected [21]. This will be the subject of future studies.

The gel adhering on a surface advances by polymerization at the leading edge and depolymerization at the rear. The motion is possible because of the rupture of the adhesion points that we just discussed. The polymerization reactions are far from equilibrium and we assume for simplicity that only polymerization occurs at the leading edge and only depolymerization occurs at the rear. The free energy gain in these reactions (as well as the energy required to maintain the *pH* gradient) are the sources of energy for the motion. The dissipation is due mainly to the gel retraction after the rupture of adhesion points.

In order to obtain a complete model for motion, we have to characterize the polymerization and depolymerization reactions. We denote the local concentration of free monomers by $\rho(x)$, the local gel concentration by $c(x) = c_0 \Lambda(x)^{-2}$, where c_0 is the concentration at the leading edge of the gel. The total number of monomers in the gel $m_{\rm gel}$ increases by polymerization and decreases by depolymerization. The kinetics of these reactions is written as

$$\frac{dm_{\text{gel}}}{dt} \Big|_{l} = h_{l}k_{p}c_{0}\rho_{l},$$

$$\frac{dm_{\text{gel}}}{dt} \Big|_{r} = -h_{r}k_{d}c_{r} = -k_{d}hc_{0}/\Lambda_{r}.$$
(4)

Here the subscripts l and r refer, respectively, to the leading edge of the gel where polymerization occurs and to the rear where the depolymerization occurs. The rate constants for polymerization k_p and depolymerization k_d should be taken at the local pH (at the leading edge for polymerization and at the rear for depolymerization). In a steady state, the gel mass is constant and $(dm_{\rm gel}/dt)|_l + (dm_{\rm gel}/dt)|_r = 0$. In the reference frame of the moving gel, the transport of the depolymerized monomers occurs by diffusion. In a steady state, the flux $J = D_m(\rho_r - \rho_l)/(L + L')$ is equal to $-(dm_{\rm gel}/dt)|_r/h$, $(D_m$ being the monomer diffusion constant). This fixes the monomer concentrations at the leading edge and at the rear $\rho_l = k_d/(k_p \Lambda_r)$ and

 $\rho_r = \rho_l \{1 + k_p c_0 (L + L')/D_m\}$. The average velocity is equal to the polymerization speed $Vc_0 = \frac{1}{h} (dm_{\rm gel}/dt)$ or

$$V = k_p \rho_l = k_d / \Lambda_r. \tag{5}$$

It is selected by the steady state polymerization or depolymerization velocity. The total length of the gel L + L' can be determined, assuming the total monomer mass m (free and in the gel) is known.

The main result of this Letter is to show that a gel can sustain a steady state motion via a treadmilling mechanism even though it adheres to the substrate. No gradient in adhesive strength is required for this to happen. The two necessary ingredients are the existence of a critical rupture force for the adhesive bonds and of a selfsustained gradient leading to a change in the gel swelling. The critical rupture force depends on the loading rate; the actual value must therefore be determined taking the gel speed into account. Furthermore, the existence of the mechanism does not depend on the sign of the induced gel swelling. The gradient that controls swelling has been described in terms of a pH gradient, but any other gradient in the gel could play this role. The differential swelling of the gel could also result from the action of molecular motors as will be discussed in a forthcoming publication. We have presented a generic mechanism. Its originality and robustness stem from the peculiar distribution of the stresses at the gel-substrate interface. Over most of the gel surface, the stress pulling on the adhesive bonds is proportional to the gel thickness and the swelling gradient. In a small region where the rupture occurs, conventional elasticity tells us that it is of opposite sign and larger by a factor L/h. For nematodes, L/h is of order 10-20 and there is a broad range of adhesive strength over which the mechanism can work. This stress distribution can be measured using techniques recently developed to study the forces that cells develop on soft substrates [22]. A sign inversion of the stress as proposed here has been observed on keratocyte cells [23].

It is very instructive to study within our framework the situation where external forces act on the moving gel. The effect of a force has a striking signature depending on the position where the force is applied. Consider an external force applied at position x along the substrate. If the force is acting at a point inside the adhering gel, the elasticity equations show that its effect is screened: the gel deformation and the stress on the surface are changed only over a region of size h (the gel thickness) around the point of application; the force therefore has no effect on the gel velocity, total length, and adhering length until it reaches a bond rupture value. If the force is applied at the front of the gel, its effect is also localized. The force, however, tends to reduce the polymerization rate; the behavior of the gel in a steady state strongly depends on the depolymerization conditions at the rear of the gel. If the force is applied at the rupture point, it must be added to the macroscopic force balance in the gel which determines the stress at the rupture point. In the presence of an additional force, the adhering section of the gel is increased and corresponds to an effective rupture force $f_c + f/(\rho_c h)$. Eventually, a force applied at the rear of the gel and opposing the motion influences the depolymerization. In general, it increases the depolymerization rate and according to Eq. (5) the velocity increases. The gel in this case has a negative mobility. All these surprising predictions could be tested experimentally. We will present a more detailed study of the effect of a force in a future publication [18].

We are grateful to J. Plastino (Institut Curie) for useful discussions on the MSP protein and A. Mogilner for interesting discussions on Ref. [7].

- [1] T. Mitchison and L. Cramer, Cell 84, 371 (1996).
- [2] T. Svitkina, A. Verkhovsky, K. McQuade, and G. Borisy, J. Cell Biol. 139, 397 (1997).
- [3] A. Verkhovsky, T. Svitkina, and G. Borisy, Curr. Biol. 9, 11 (1999).
- [4] P.G. DeGennes, C. R. Acad. Sci. 323, 663 (1996).
- [5] F. Gerbal, P. Chaikin, Y. Rabin, and J. Prost, Biophys. J. 79, 2259 (2000).
- [6] A. Bernheim-Groswasser, S. Wiesner, R. Goldsteyn, M. F. Carlier, and C. Sykes, Nature (London) 417, 308 (2002).
- [7] D. Bottino, A. Mogilner, T. Roberts, M. Stewart, and
 G. Oster, J. Cell Sci. 115, 367 (2002); see also
 A. Molginer and D. Verzi, J. Stat. Phys. (to be published).
- [8] J. Theriot, Cell 84, 1 (1996).
- [9] T. Roberts and M. Stewart, J. Cell Biol. 149, 7 (2000).
- [10] K. King, M. Stewart, and T. Roberts, J. Cell Sci. 107, 2941 (1994).
- [11] J. Italiano, T. Roberts, M. Stewart, and C. Fontana, Cell 84, 105 (1994).
- [12] T. Roberts, E. Salmon, and M. Stewart, J. Cell Biol. 140, 367 (1998).
- [13] J. Italiano, M. Stewart, and T. Roberts, J. Cell Biol. 146, 1087 (1999).
- [14] K. King, J. Essig, T. Roberts, and T. Moerland, Cell Motil. Cytoskeleton 27, 193 (1994).
- [15] For a simple introduction to the elasticity of continuous media, see, for example, P. Chaikin and T. Lubensky, *Principles of Condensed Matter Physics* (Cambridge University Press, Cambridge, 1995).
- [16] L. Landau and E. Lifschitz, *The Theory of Elasticity* (Butterworth-Heinemann, Boston, 1995).
- [17] G. Batchelor, An Introduction to Fluid Dynamics (Cambridge University Press, Cambridge, 1967).
- [18] P. Dewimille *et al.* (to be published).
- [19] G. Bell, Science 200, 618 (1978).
- [20] E. Evans and K. Ritchie, Biophys. J. 72, 1451 (1997).
- [21] T. Baumberger, C. Caroli, and O. Ronsin, Phys. Rev. Lett. 88, 075509 (2002).
- [22] N. Balaban, U. Schwarz, D. Riveline, P. Goichberg, G. Tzur, I. Sabanay, D. Mahalu, S. Safran, A. Bershadsky, L. Addadi, and B. Geiger, Nature Cell Biol. 3, 466 (2001).
- [23] C. Galbraith and M. Sheetz, J. Cell Biol. 147, 1313 (1999).

168102-4 168102-4