

Beyond Molecular Dynamics: Long Time Atomic-Scale Simulations – Dresden, March 26 - 29, 2012

# Monte Carlo simulations of diffusive phase transformations: time-scale problems

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# Outline

- § Atomistic Kinetic Monte Carlo simulations (AKMC) : from an atomistic description of diffusion mechanisms to the kinetics of phases transformations in metallic alloys
- § The precipitation kinetic pathways depends on point defect diffusion properties
  - key points : the dependence of the *migration barriers* and the *vacancy concentrations* with the local configuration

*Ab initio* calculations (thermodynamics and diffusion)



Diffusion model on a rigid lattice



Atomistic Kinetic Monte Carlo simulations (AKMC)

- § Applications to the decomposition of Fe-Cr concentrated solid solutions  
Comparison with experiments (3D atom probe and SANS)
- § Contributions of *ab initio* calculations (and their current limitations)

# Rigid lattice diffusion model

- A-B alloy – rigid lattice with pair interactions :  $\epsilon_{AA}$ ,  $\epsilon_{AB}$ ,  $\epsilon_{BB}$  +  $\epsilon_{AV}$ ,  $\epsilon_{BV}$

Possible improvements:

- many-body interactions (triangle, tetrahedron, etc.)
- composition dependent interactions (-> FeCr alloys)
- temperature dependent interactions (->  $\Delta S_{\text{mig}}$ ,  $\Delta S_{\text{for}}$ )

Alternative approach : interatomic potentials

- Diffusion by thermally activated point defects jumps

- jump frequency : depend on the local environments

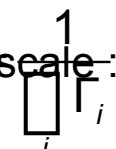
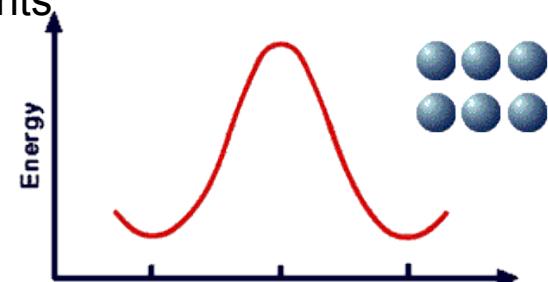
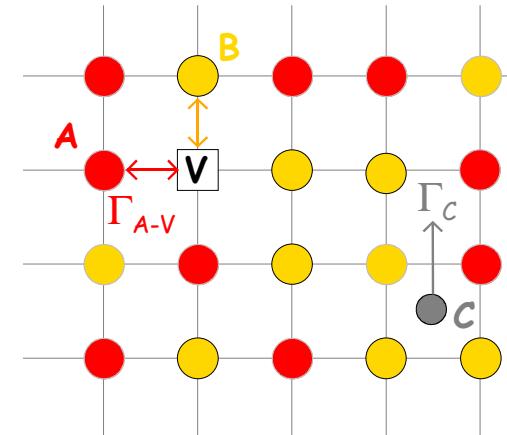
$$\Gamma_{AV} = v_A \exp \left( -\frac{\Delta E_{AV}^{\text{mig}}}{k_b T} \right)$$

- migration barriers: broken-bond models

$$\Delta E_{AV}^{\text{mig}} = E_{\text{sys}}(\text{SP}) - E_{\text{sys}}(\text{ini}) = \underbrace{\epsilon_{Ai}^{\text{SP}}}_{\text{saddle-point interactions}} - \underbrace{\epsilon_{Aj}^{(n)} - \epsilon_{Ak}^{(n)}}_{\text{broken-bonds}}$$

- AKMC: residence time algorithm

Simulations with 106-107 atoms, PBC and 1 vacancy -> time scale:



# Physical time scale and point defect concentrations

The kinetic pathways depends on the migration barriers but also on the point defect concentrations : simulations with a constant number of point defects (e.g.  $NV = 1$ ) require a time rescaling

The time correction factor *is not constant*

$$t = t_{MC} \frac{\langle C_V^{MC} \rangle}{\langle C_V^{eq} \rangle} \text{ with } \langle C_V^{MC} \rangle = N_V / N$$

$\langle C_V^{eq} \rangle$  usually evolves during the phase transformations

- For any local environment ( $\alpha$ )  $t = t_{MC} \frac{C_V^{MC}(\alpha)}{C_V^{eq}(\alpha)}$  with  $C_V^{eq}(\alpha) = \exp \left( -\frac{E_V^{for}(\alpha)}{kT} \right)$
- also provides an estimation of  $\langle C_V^{eq} \rangle = \langle C_V^{MC} \rangle \frac{C_V^{eq}(\alpha)}{C_V^{MC}(\alpha)}$  during the phase transformation
- A convenient choice for phase separation : pure A or pure B  

$$E_V^{for}(\text{pure A}) \approx -\frac{Z}{2} \xi_A + Z_{AV}$$
- Only valid when vacancy concentration remains at equilibrium.  
Alternative approach : AKMC with formation and annihilation mechanisms (sources/sinks)

# One simple example

Phase separation in an A95-B5 alloy :  $E_V^{for}(A) = 1.4 \text{ eV} > E_V^{for}(B) = 1.0 \text{ eV}$

AKMC simulation  
with 1 vacancy  
 $T = 573 \text{ K} (0.6 \text{ Tc})$

$$\langle C_V^{eq} \rangle = \langle C_V^{MC} \rangle \frac{C_V^{eq}(A)}{C_V^{MC}(A)} = \langle C_V^{MC} \rangle \frac{C_V^{eq}(B)}{C_V^{MC}(B)}$$

$$\langle C_V^{eq} \rangle$$

Strong variation of the time rescaling factor

Gibbs-Thomson effect :

Application: vacancy concentration in non-ideal concentrated solid solutions  
(Mean-Field models, M. Nastar)

$\langle C_V^{eq} \rangle$  with ref. A slightly  $>_{\%} \langle C_V^{eq} \rangle$  with ref. B

# Copper precipitation in $\alpha$ -iron : clusters mobility

$$E_V^{for}(\text{Cu}) = 0.9 \text{ eV} > E_V^{for}(\text{Fe}) = 2.1 \text{ eV}$$

Classical theories of nucleation, growth and coarsening: emission/absorption of individual solute atoms

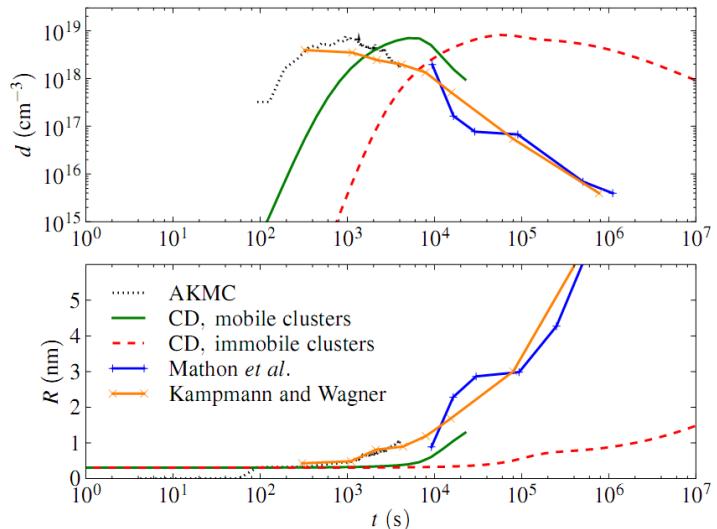
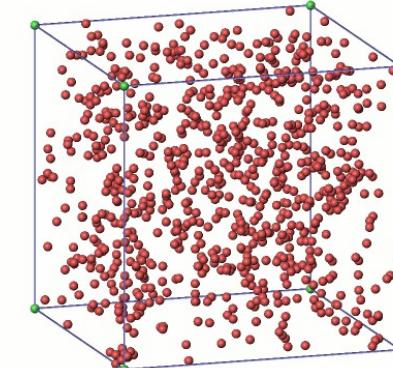
## · AKMC simulation :

Small Cu clusters are mobile than monomers  
-> coagulation mechanisms

## · Measurements of the cluster diffusion coefficients

### · CD and EKMC simulations (T. Jourdan)

- the diffusion of clusters strongly accelerates the precipitation kinetics
- better agreement with experiments



# $\alpha$ - $\alpha'$ decomposition in Fe-Cr alloys

Fe-Cr alloys : a model system for ferritic-martensitic steels

A special thermodynamic behavior :

SANS and electrical resistivity experiments:

below 10%Cr : ordering tendency

above 10%Cr : unmixing tendency

(Mirebeau et al, 1984)

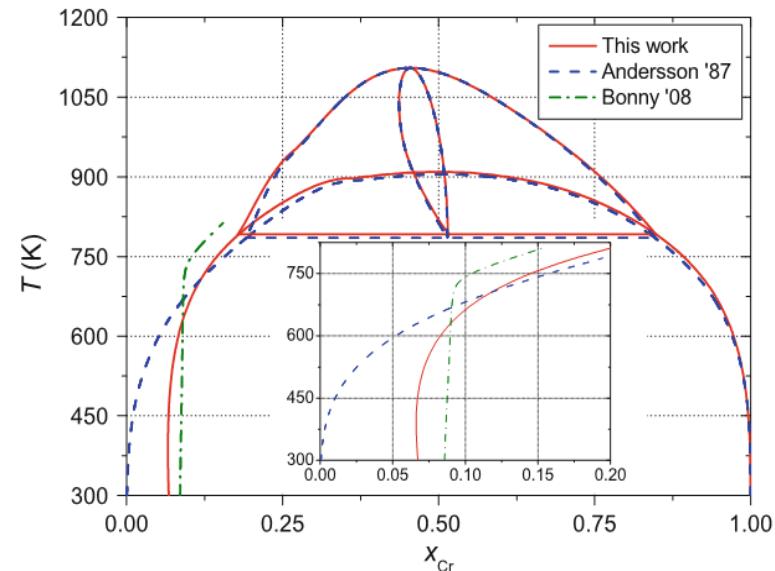
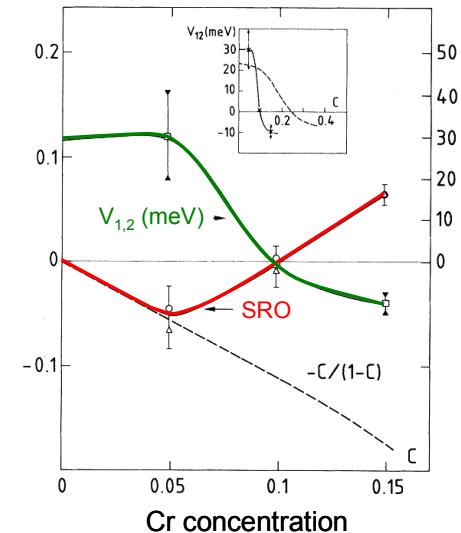
*Ab initio* calculations: connected to magnetic properties

Strong vibrational entropy (exp. + *ab initio*)

Phase diagram

“old Calphad”: no Cr solubility at low

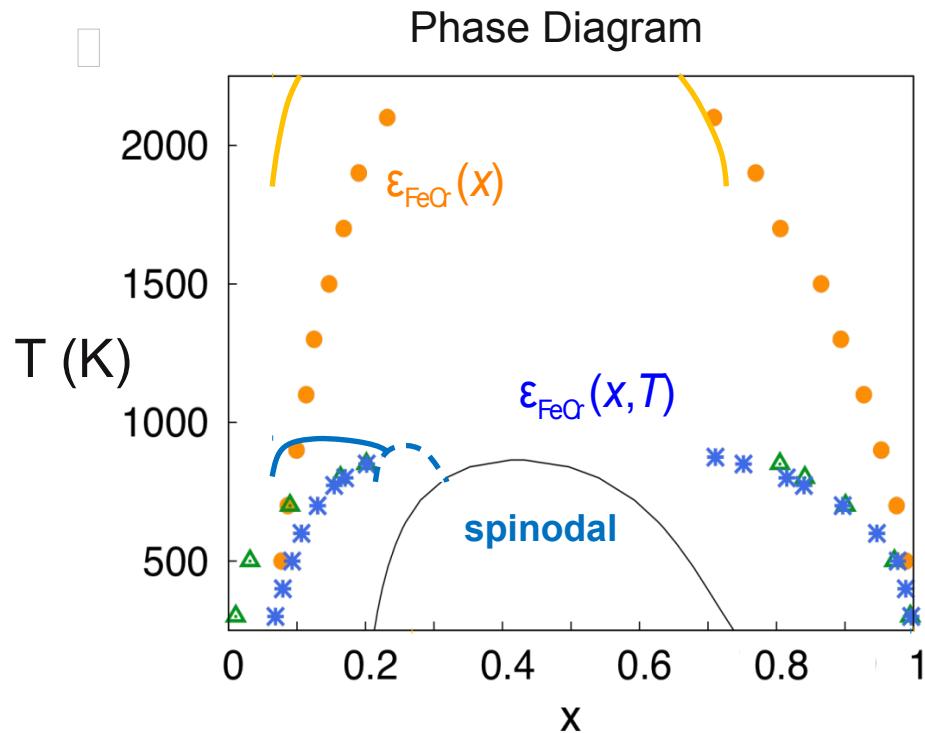
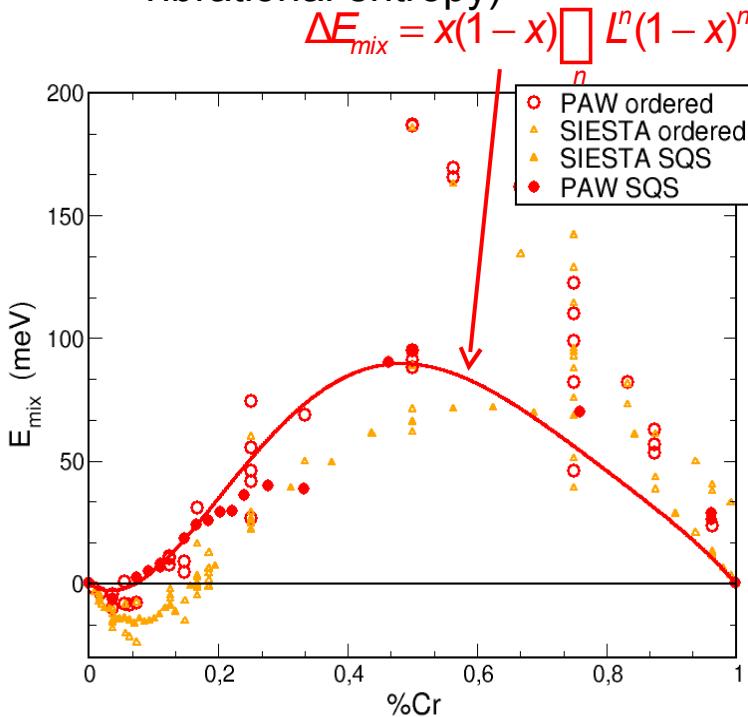
new phase diagram  
takes into account DFT calculations  
and experiments at low T  
(Bonny et al, 2010)



# Thermodynamics: pair interaction model

(M. Levesque et al. 2010)

- Constant pair interactions:  $\Delta E_{mix} = x(1-x)\Omega$  with  $\Omega = \frac{1}{n} \sum_n \left( \epsilon_{AA}^n + \epsilon_{BB}^n - 2\epsilon_{AB}^n \right)$  & symmetrical mixing energies and phase diagram
- Composition dependent pair interactions fitted on DFT calculations (1st and 2nd nearest-neighbor) -> asymmetrical phase diagram
- Temperature dependence:  $\epsilon_{FeCr}(x, T) = \epsilon_{FeCr}(x, 0) (1-T/\theta)$  (magnetic effects, vibrational entropy)



# Diffusion properties: migration barriers

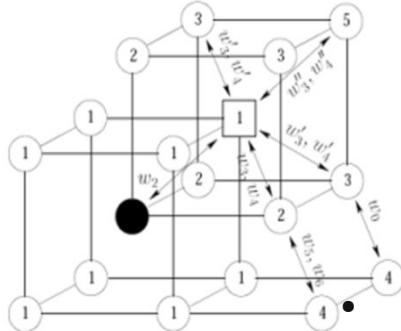
- Vacancy migration barriers : *vacancies-atom* and *saddle-point* (SP) pair interactions fitted on DFT calculations of vacancy formation energies and migration barriers (DFT-SIESTA)

attempt frequencies  $v_{Fe}$  and  $v_{Cr}$  fitted on *ab initio* or the experimental pre-exponential factors

- Self  $D_A$  diffusion in pure metals

- Diffusion in dilute alloys :

$$D_B^{(10\text{-frequency model})} = a \frac{c_V(A)}{\prod_{i=1}^2 w_i}$$



	Fe	Cr
$E_V^{\text{for}}$	2.18	1.91 (Spin-wave)
$E_V^{\text{mig}}$	0.69	1.25 (AFM configuration)
Q	2.87	3.16
exp:	2.91	exp: 3.2-3.6 (at low T)

Activation energy at 0 K (eV)	in Fe		in Cr	
	SIESTA	RLA	SIESTA	RLA
$\Delta H_0$	0.69	0.69	1.25	1.25
$\Delta H_2$	0.57	0.57	1.16	1.16
$\Delta H_3$	0.64	0.80	1.04	1.12
$\Delta H'_3$	0.69	0.70	1.10	1.24
$\Delta H''_3$	0.67	0.70	1.07	1.24
$\Delta H_4$	0.66	0.79	1.14	1.12
$\Delta H'_4$	0.65	0.67	1.31	1.25
$\Delta H''_4$	0.66	0.67	1.27	1.25
$\Delta H_5$	0.74	0.69	1.32	1.25
$\Delta H_6$	0.74	0.68	1.20	1.25



Migration barriers are computed at 0K, in magnetic configurations  
extrapolation above the transitions temperature ?

# Diffusion coefficients

Pre-exponential factor :  $v_D = 10^{13} \text{ s}^{-1}$

$S_V^{for}(\text{Fe}) = 4.1k_B$  (DFT, Lucas & Schaüblin, 2009)

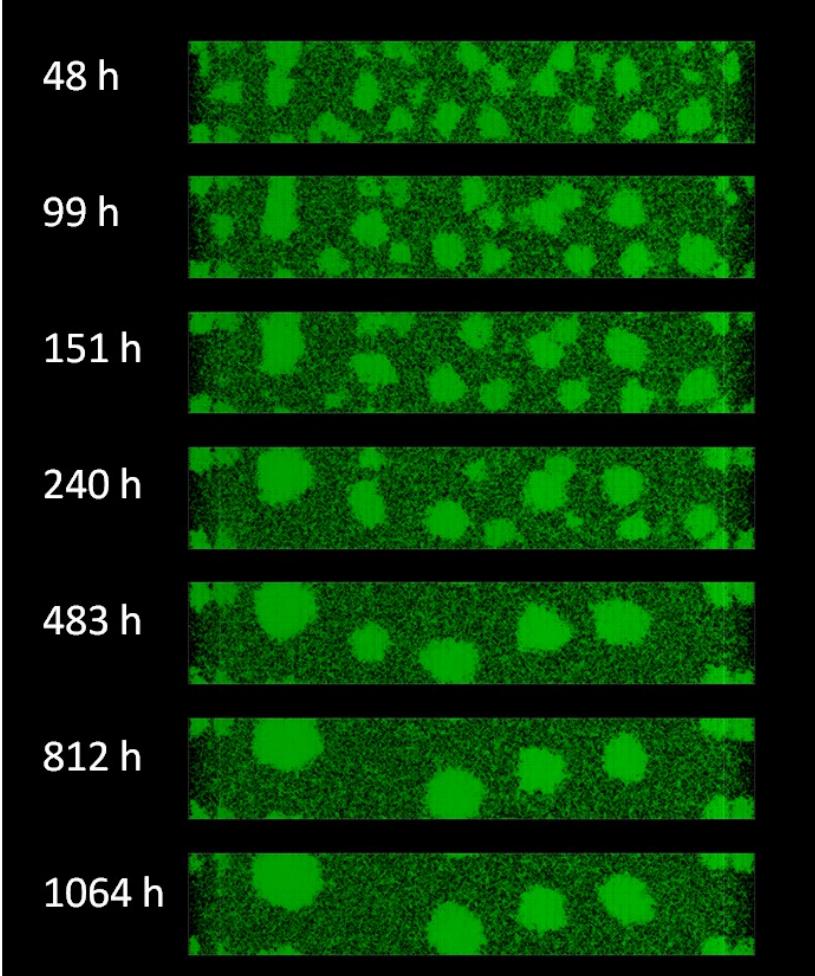
$S_V^{mig}(\text{Fe}) = 2.1k_B$  (Athènes and Marinica, 2010)

- Good agreement with the experimental  $D_0$  (in iron)
  - in iron
  - in chromium

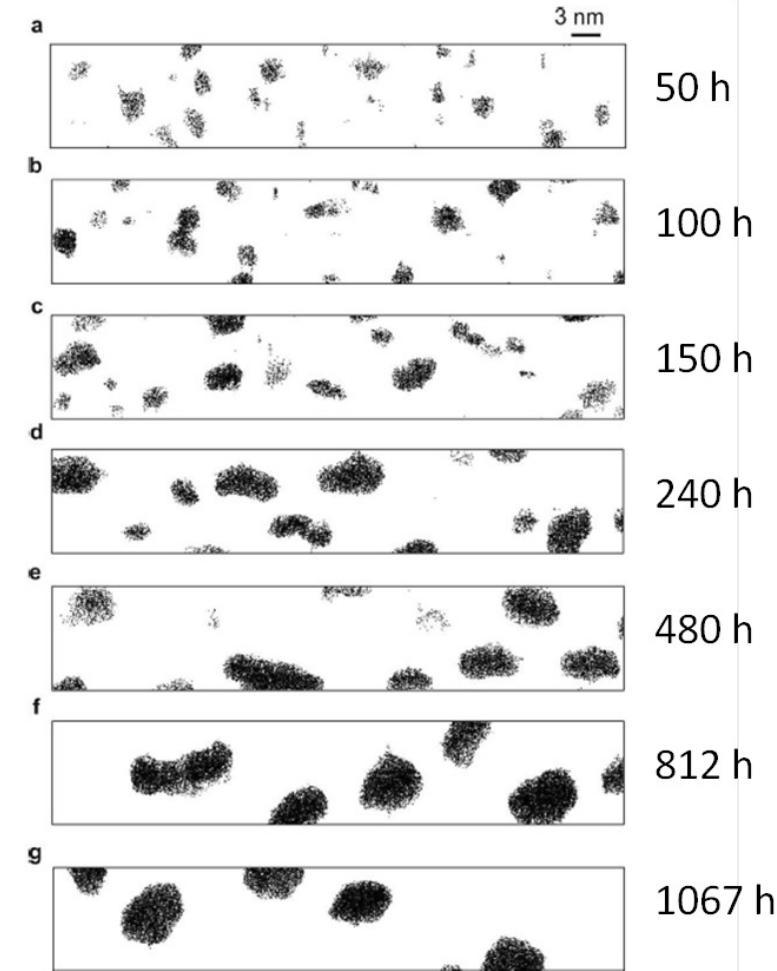
# AKMC: $\alpha$ - $\alpha'$ decomposition during thermal ageing

Fe-20%Cr   T = 500°C

AKMC (E. Martinez, O. Senninger, CEA)  
Rouen, 2009



3D atom probe (Novy et al, GPM



# AKMC vs SANS experiments

(E. Martinez, O. Senninger, 2011)

Small Angle Neutrons Scattering experiments : Bley 1994, Furusaka et al 1986

energie atomique + énergies alternatives

Fe – 20, 35 and 50% Cr, 500°C: good agreement with SANS experiments

Fe – 40%Cr, 540°C : AKMC much slower than SANS experiments

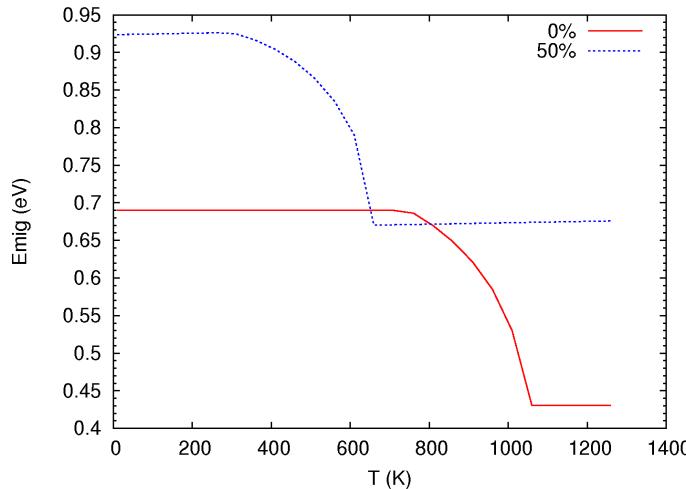
The Curie temperature decreases with the Cr content: effect of the ferro-paramagnetic transition?

# Effect of the magnetic transition on the kinetics

O. Senninger, E. Martinez et al. 2012

Evolution of the Curie temperature  
with the composition

Decrease of the migration barriers at TC (fitted on the experimental diffusion coefficients)



# Radiation induced segregation and precipitation

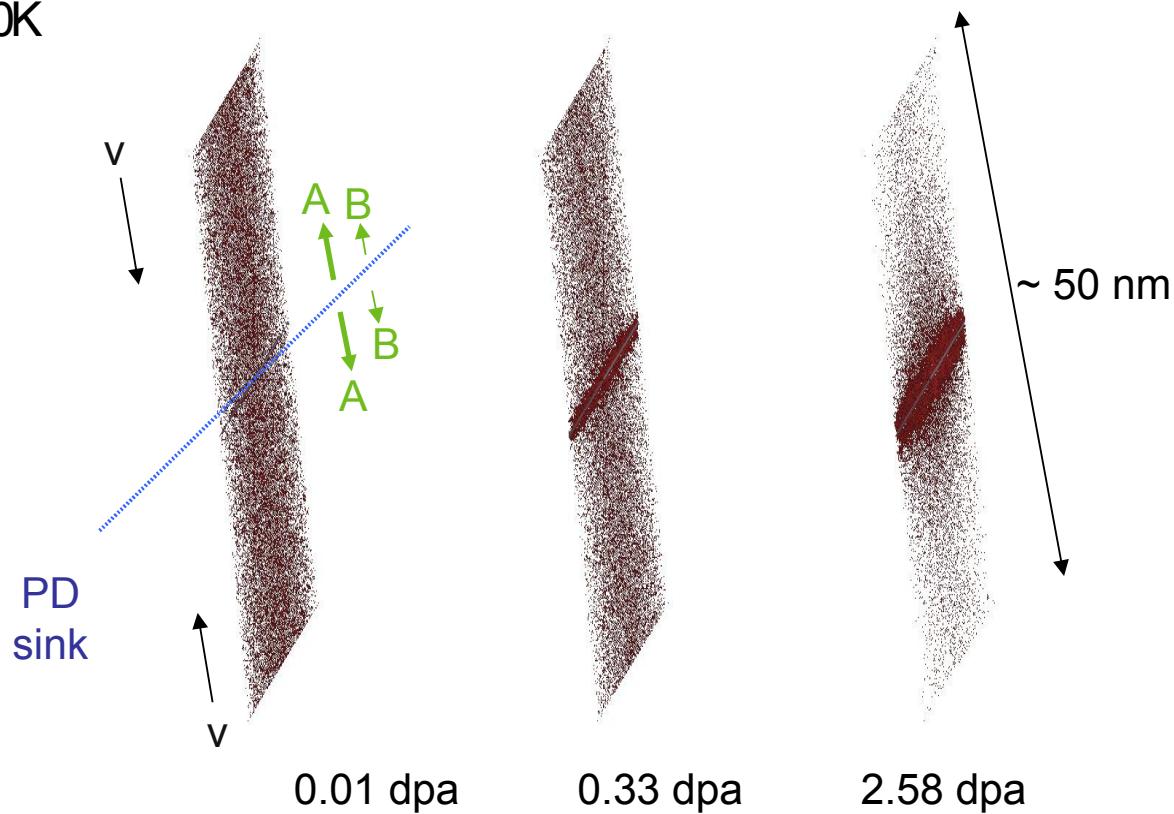
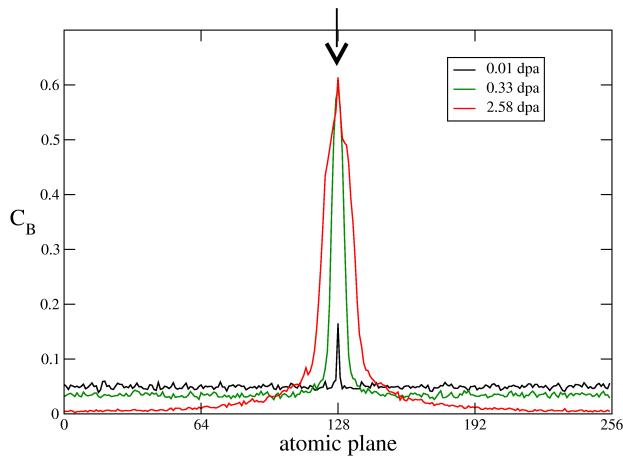
Undersaturated solid solution with an unmixing tendency ( $\varepsilon_{AA} + \varepsilon_{BB} - 2\varepsilon_{AB} < 0$ )

$$C_B = 5\% - C_B^{eq} = 8\%$$

$$D_B^V / D_A^V = 0.06 \text{ and } D_B^I / D_A^I = 1$$

$$G = 10^{-6} \text{ dpa.s}^{-1} \quad T = 800\text{K}$$

Solute concentration profile  
Point defect sink



# Conclusions

AKMC simulations: a detailed description of thermodynamic and diffusion properties

- dependence of point defect jump frequencies and concentrations with the local atomic environment
- correlation effects ( $\leftrightarrow$  theory of diffusion in alloys)

But time consuming:  $\rightarrow$  cluster dynamics, OKMC, EKMC

*Ab initio* calculations :  $\rightarrow$  reliable parameters at 0K

temperature effects : vibrational entropy of mixing, vacancy formation and migrations entropies

- in pure metals

-

- in concentrated alloys ?

Fe-Cr alloys: importance of magnetic transitions

Perspectives : radiation induced segregation, spinodal decomposition in Fe-Cr alloys