

Density-fitted periodic local MP2. Application to molecular crystals and surface adsorption

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CRYSCOR overview I

Provides a periodic local MP2 solutions on top of the HF reference, provided by the CRYSTAL code.

CRYSCOR takes (from CRYSTAL):

- The information on the **basis set, geometry, symmetry, etc**
- **Density and Fock matrices** of the **periodic HF** solution
- Symmetrized **Wannier functions**

CRYSCOR delivers:

- **periodic LMP2 energy**
- **periodic LMP2 density matrices** (presently without an orbital relaxation)

CRYSCOR overview II

The DFP-LMP2 (**density fitted periodic local MP2**) method comprises:

- Molecular local correlation techniques: **PAOs for virtuals, excitation domains, orbital pairs**
- Extensive use of **symmetry**
- Two-step multipole-constrained **Density Fitting** approximation
- Distant-pair two-electron integrals – via **multipole expansion**
- The **contribution from the very distant pairs** – through evaluation of C_6 parameters

The CRYSCOR crew

Torino

Regensburg

Prof. Cesare Pisani

Prof. Martin Schütz

Dr. Lorenzo Maschio

Dr. Denis Usvyat

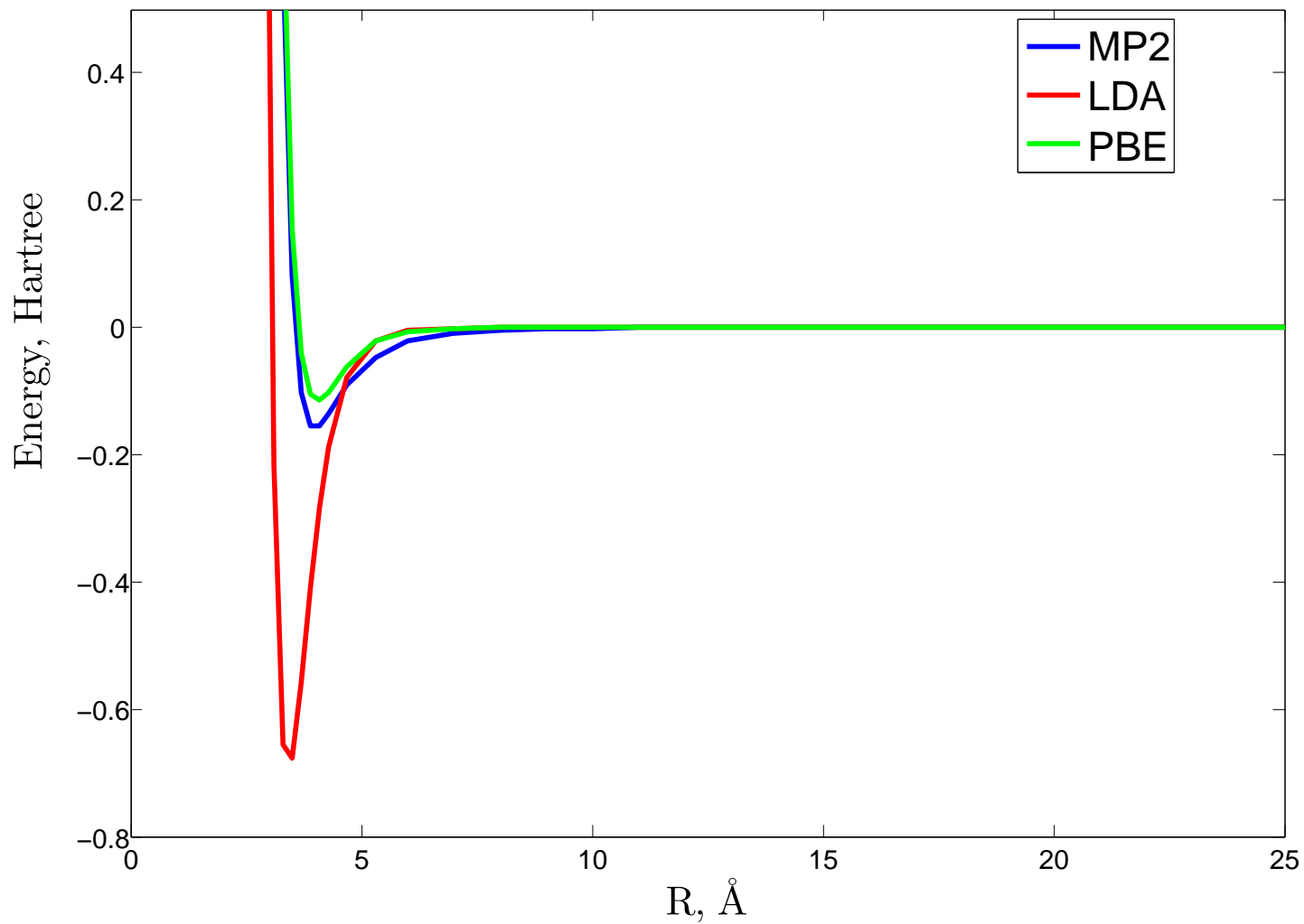
Dr. Silvia Casassa

Migen Halo

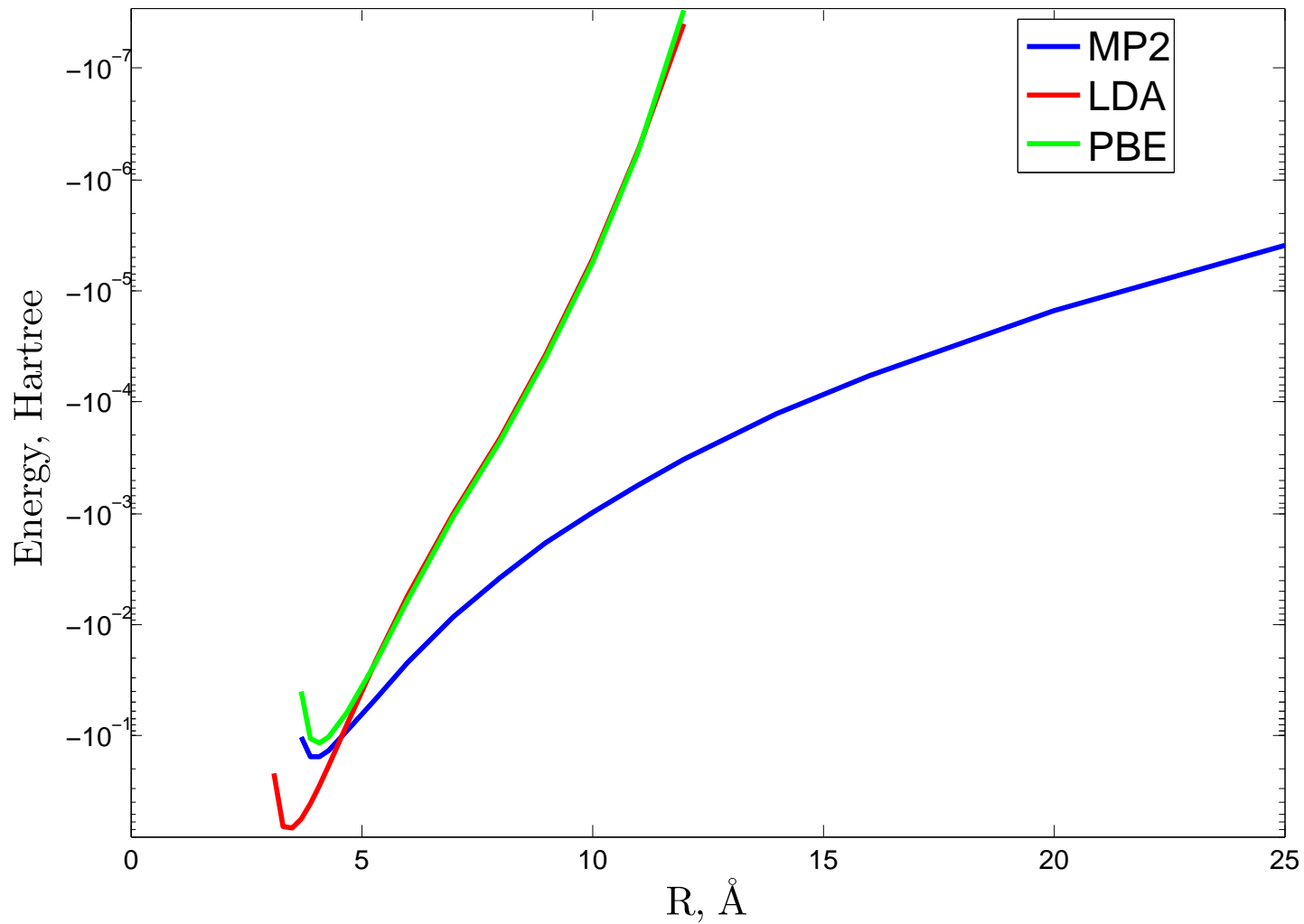
LMP2 and weakly bound systems

- **Unlike DFT LMP2 is able to treat dispersion**, which plays an important role in such systems!
- **Wannier functions are well localized** within the monomers \Rightarrow the periodic LMP2 calculations are relatively not expensive.
- **However**, for describing the weak binding **large basis sets are needed**.

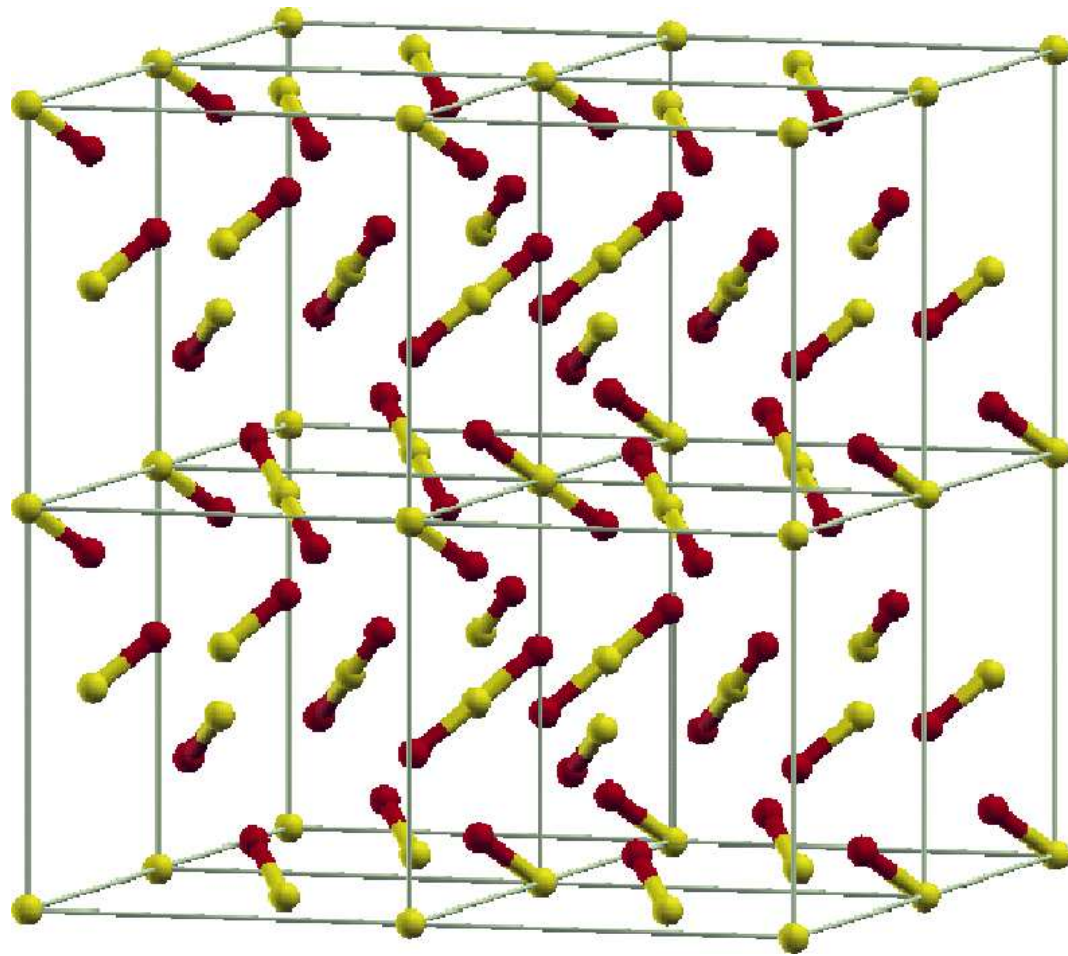
DFT and dispersion, argon dimer



DFT and dispersion, argon dimer



CO₂ crystal

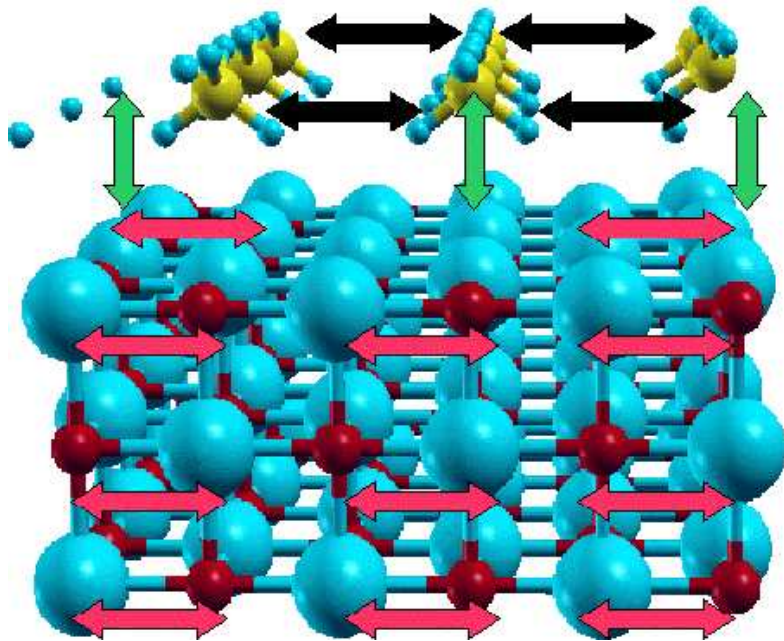


CO₂ crystal

Method	Formation energy kcal mol ⁻¹	Lattice constant Å
HF, 6-21G*	2.59	6.09
HF, 6-311G(3D)	1.95	6.13
LDA, 6-21G*	7.46	5.26
LDA, 6-311G(3D)	8.48	5.28
B3LYP, 6-21G*	1.89	6.01
B3LYP, 6-311G(3D)	1.40	6.01
LMP2, 6-21G*	4.51	5.79
LMP2, 6-21G* → AVQZ	5.95	5.65
LMP2, 6-311G(3D)	5.76	5.68
LMP2, 6-311G(3D) → AVQZ	6.50	5.63
Experiment	6.44	5.54

Physisorption on a surface.

Pair Partitioning



• Adsorbate - adsorbate pairs

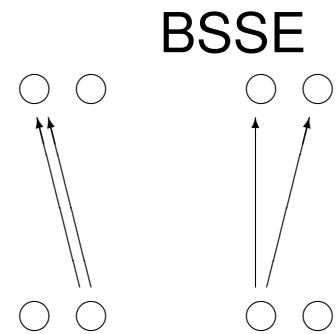
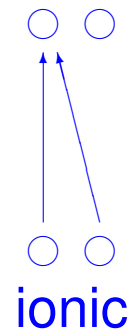
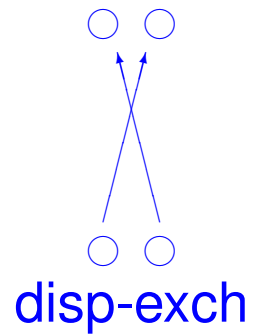
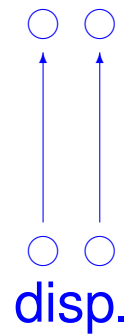
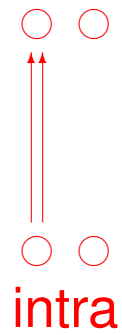
• Adsorbate - slab pairs

• Within-slab pairs

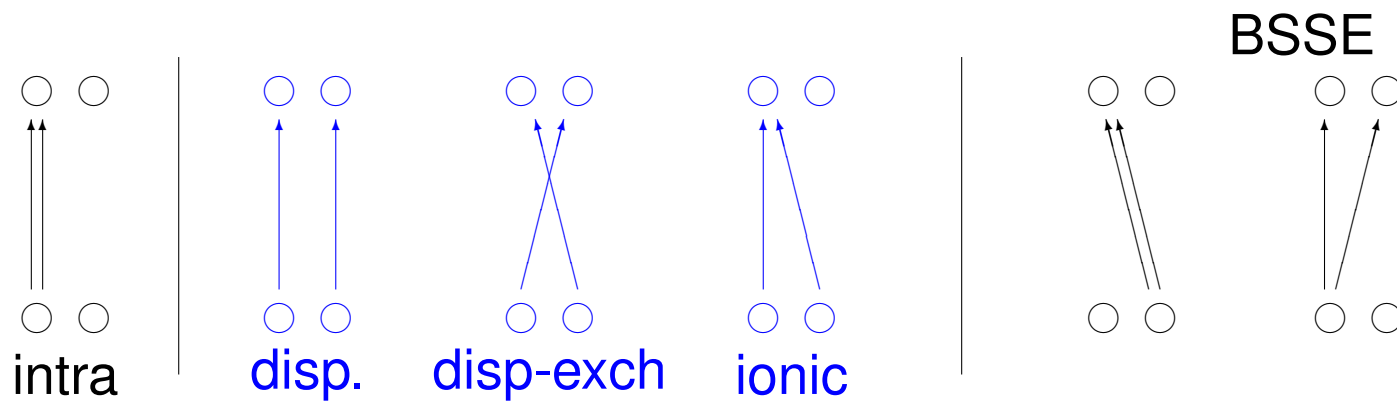
New approximation: Only important pairs can be included in the calculations.

For example, **only slab-adsorbate pairs** can be taken into account.

What we lose:



What we lose:



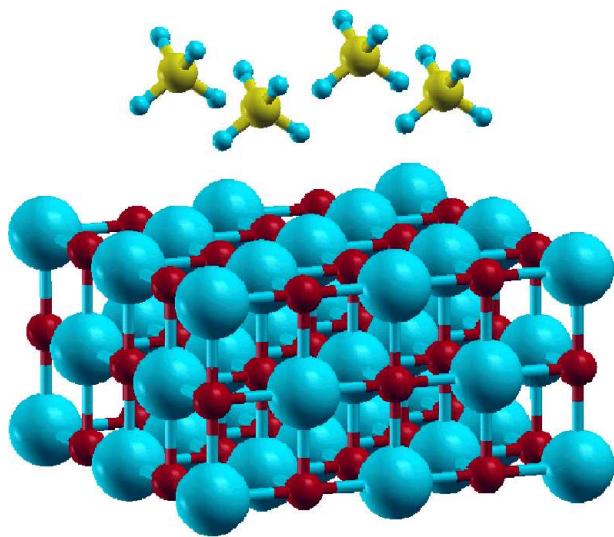
The electrostatic interaction remains at the HF level.

What we gain:

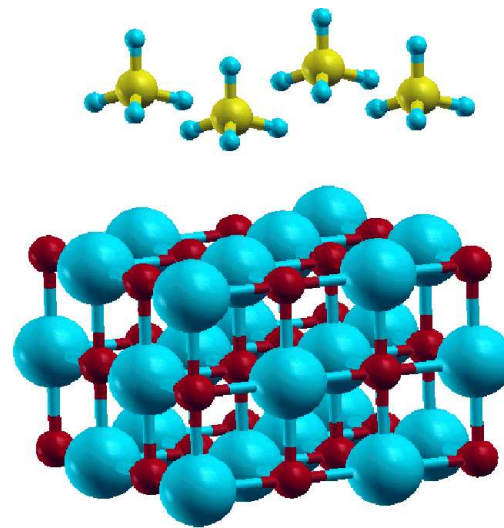
- **If electrostatic interaction doesn't dominate**, the approximation is fairly good. For adsorption of methane on MgO its error is ≈ 0.5 kJ/mol.
- **This error can be reduced further** by including just few adsorbate-adsorbate and within-slab pairs.
- The calculations become **essentially less consuming** without a large loss in accuracy.
- **No LMP2 BSSE** by construction.
- Possibility to treat molecular **adsorption on metallic surfaces**.

Methane on MgO

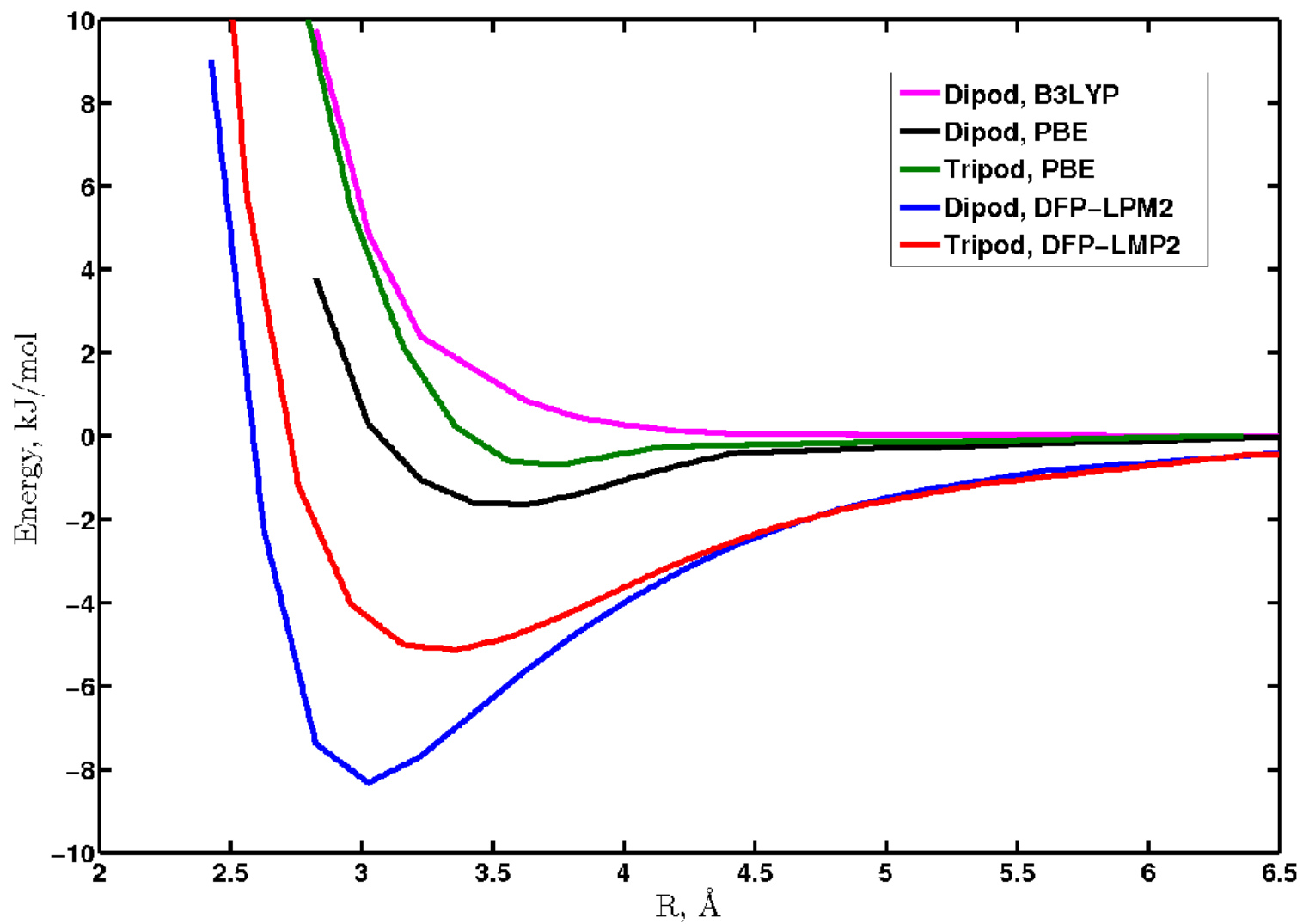
Dipod



Tripod



Methane on MgO



Methane on MgO

Method	Preferable structure	Adsorption energy, kJ/mol
Semiempirical	Tripod	
PCI-extrapolated MCPF Methane on MgO cluster*	Dipod	8
DFT**	Dipod	
DFP-LMP2, 8-511G*(MgO)/6-311G(3d)(CH₄)	Dipod	10.1
Experiment	Dipod	13.1

*K. Todnem, K. J. Borve, M. Nygren, *Surf. Sci.* 421, 296 (1999)

**M. L. Drummond, B. G. Sumpter, W. A. Shelton, and J. Z. Larese, *Phys. Rev. B* 73, 195313 (2006)

Physisorption on a metallic surface

- HF is bad for metals
- MP2 is totally inappropriate for metals

But

- We are not interested in the metal but rather in the adsorption on it.
- DFT for dispersion is worse than HF for a metal.
- Adsorbate layer does not conduct

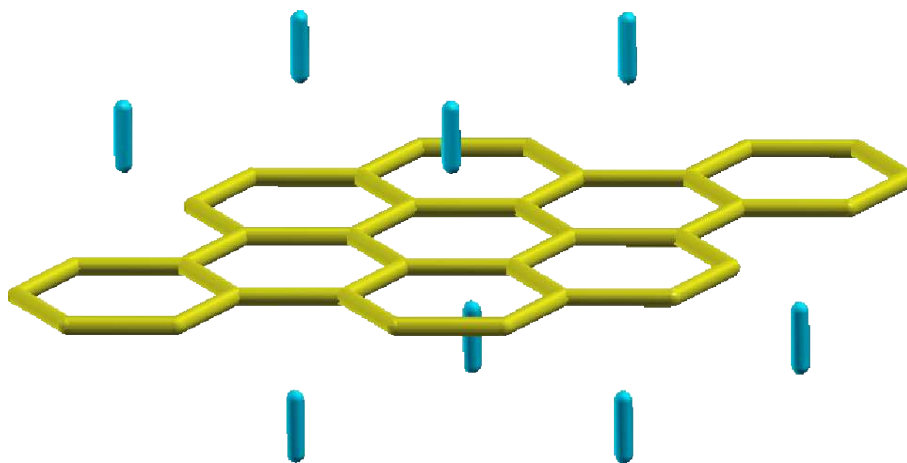
Therefore

We can try to model the dispersion by MP2 by considering only the slab-adsorbate (and if needed adsorbate-adsorbate pairs)

H₂ on graphene

H₂ physisorbed on graphene.

(A common work with together with Thomas Heine and Lyuben Zhechkov in the context of hydrogen storage)

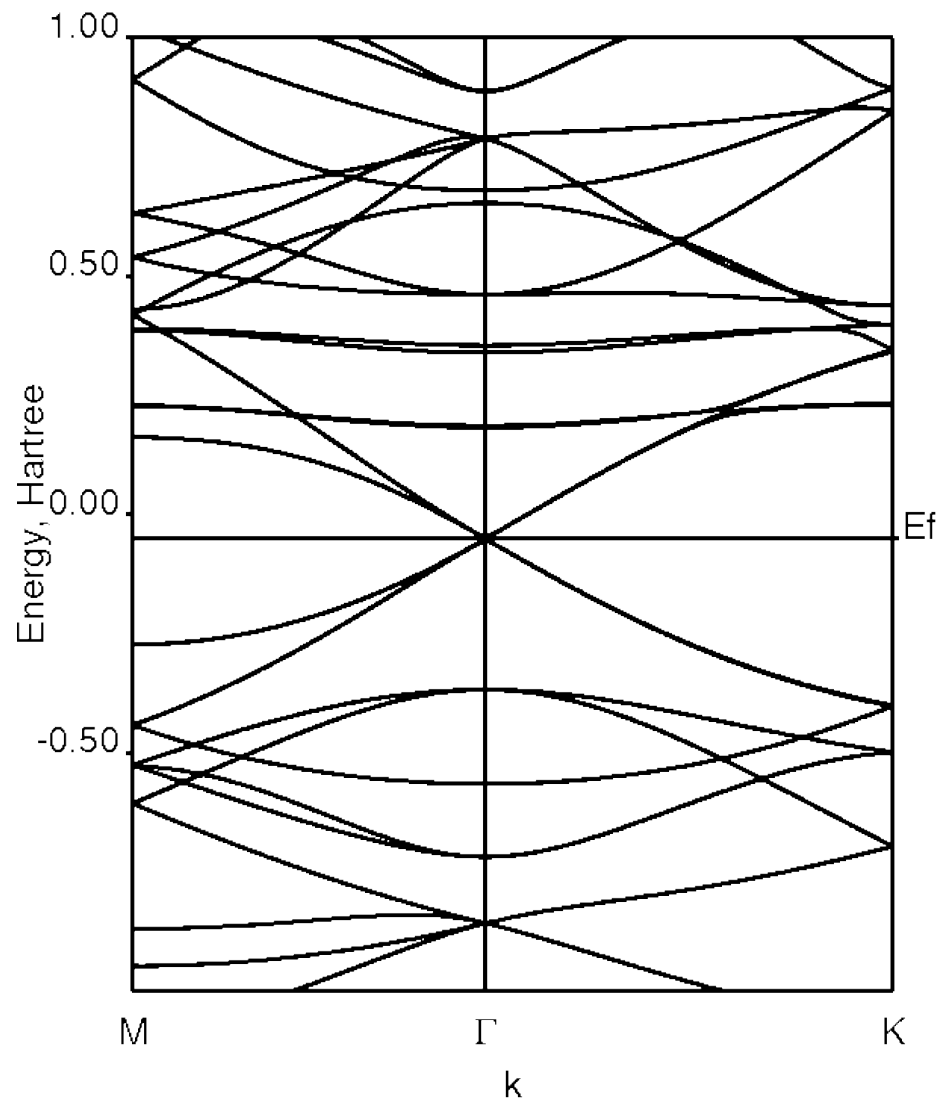


R.H. Lochan and M. Head-Gordon, PCCP, 8, 2006

H₂ on graphene

Graphene is a metal (semimetal)...

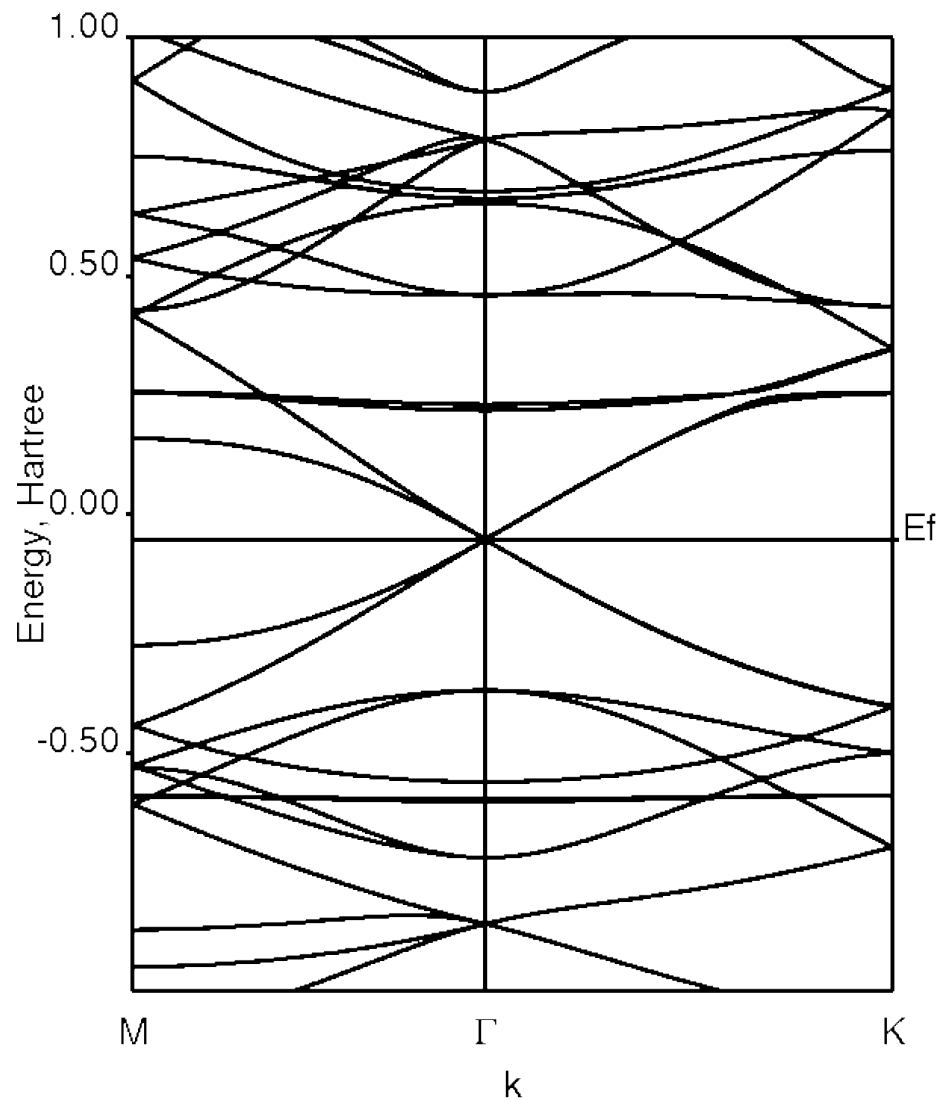
Band structure of graphene



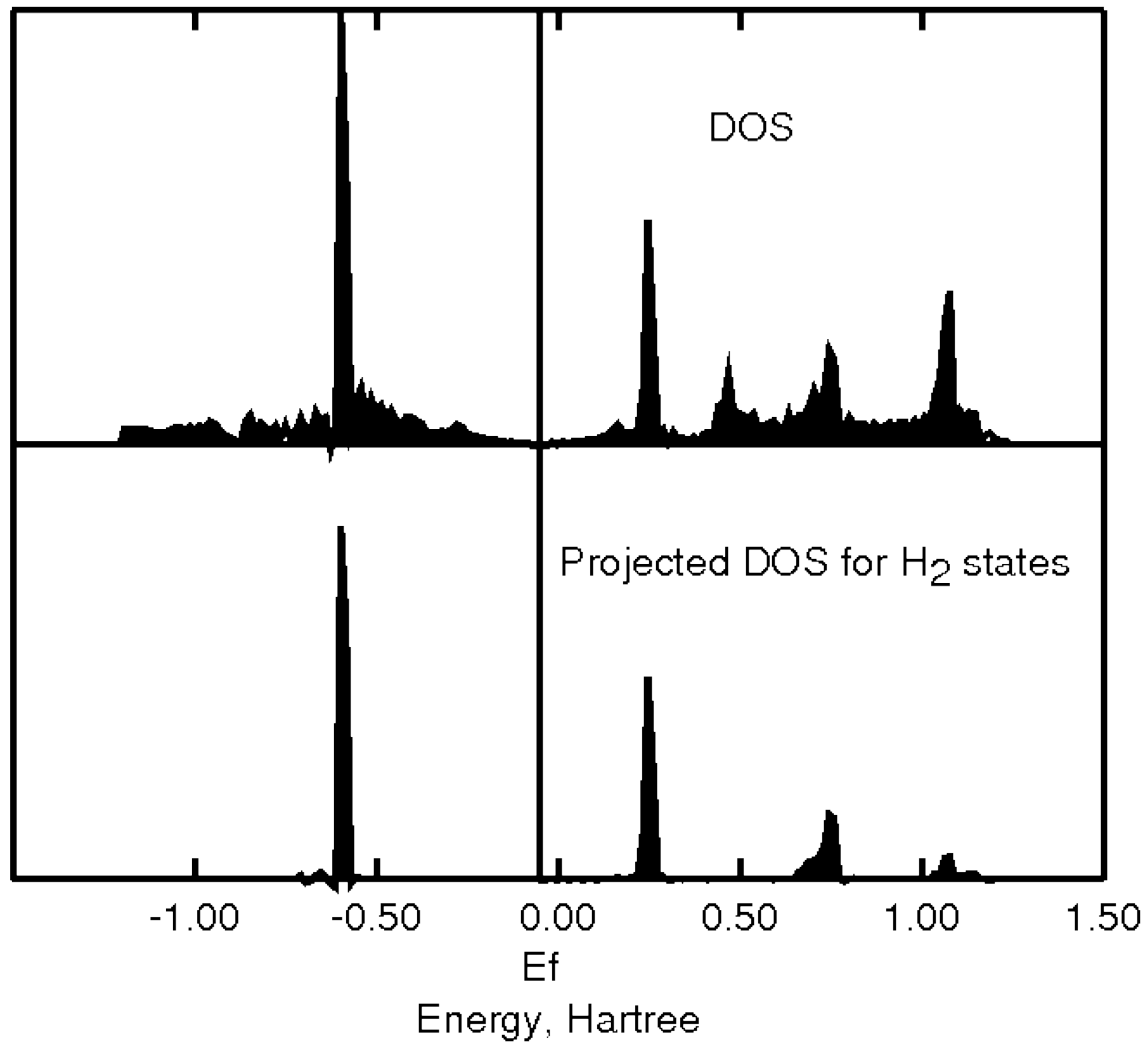
H₂ on graphene

But the states of the adsorbed H_2 molecules form **narrow bands** with a **band gap**!

Band structure of H₂ on graphene



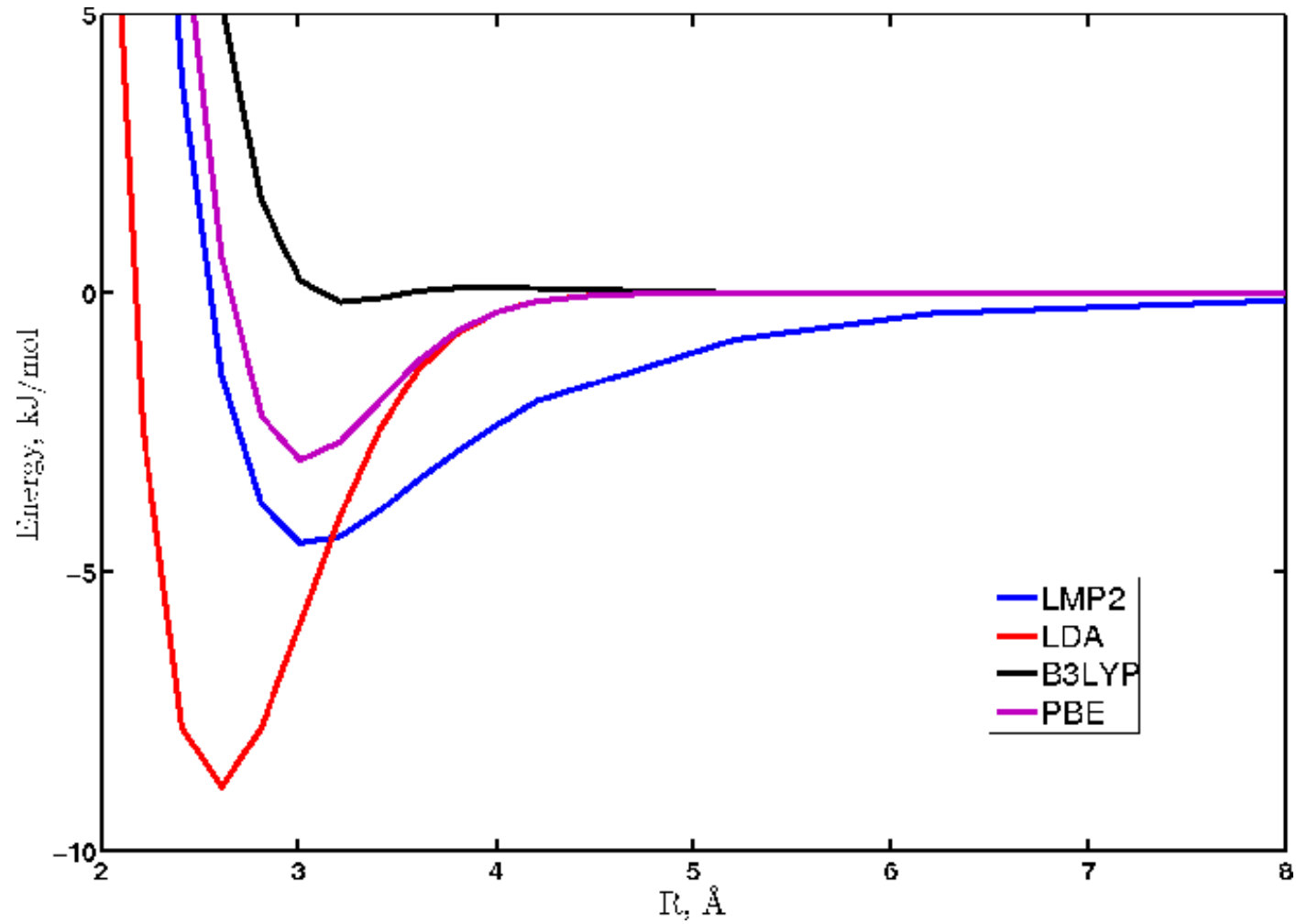
DOS of H₂ on graphene



H₂ on graphene

So the **LMP2 method can be applied** for
H₂ physisorbed on graphene
if **within-slab pairs are not included.**

H₂ on graphene



Adsorption energies for H₂ on graphene and related structures

System	Method	Adsorption energy, kJ/mol
MP2/6-31G	H ₂ on benzene	1
*MP2/6-311G**	H ₂ on benzene	2.1
*MP2/aug-cc-pVQZ	H ₂ on benzene	5
*MP2/aug-cc-pVQZ	H ₂ on coronene	6.4
Experiment	H ₂ on graphene	5
LMP2/6-31G*(C)/6-311G**(H)	H₂ on graphene	2.4

* T. Heine, L. Zhechkov, G. Seifert, *PCCP* **6** (2004)

Timings for DFP-LMP2 calculations

System and basis set	CPU time for HF, sec	CPU time for DFP-LMP2, sec
CO ₂ crystal, 6-21G*	30	3500
CO ₂ crystal, 6-311G(3d)	300	17000
CH ₄ on MgO, 6-311G(3d)/8-511G*	4500	9500
H ₂ on graphene, 6-311G**/6-31G*	700	5000

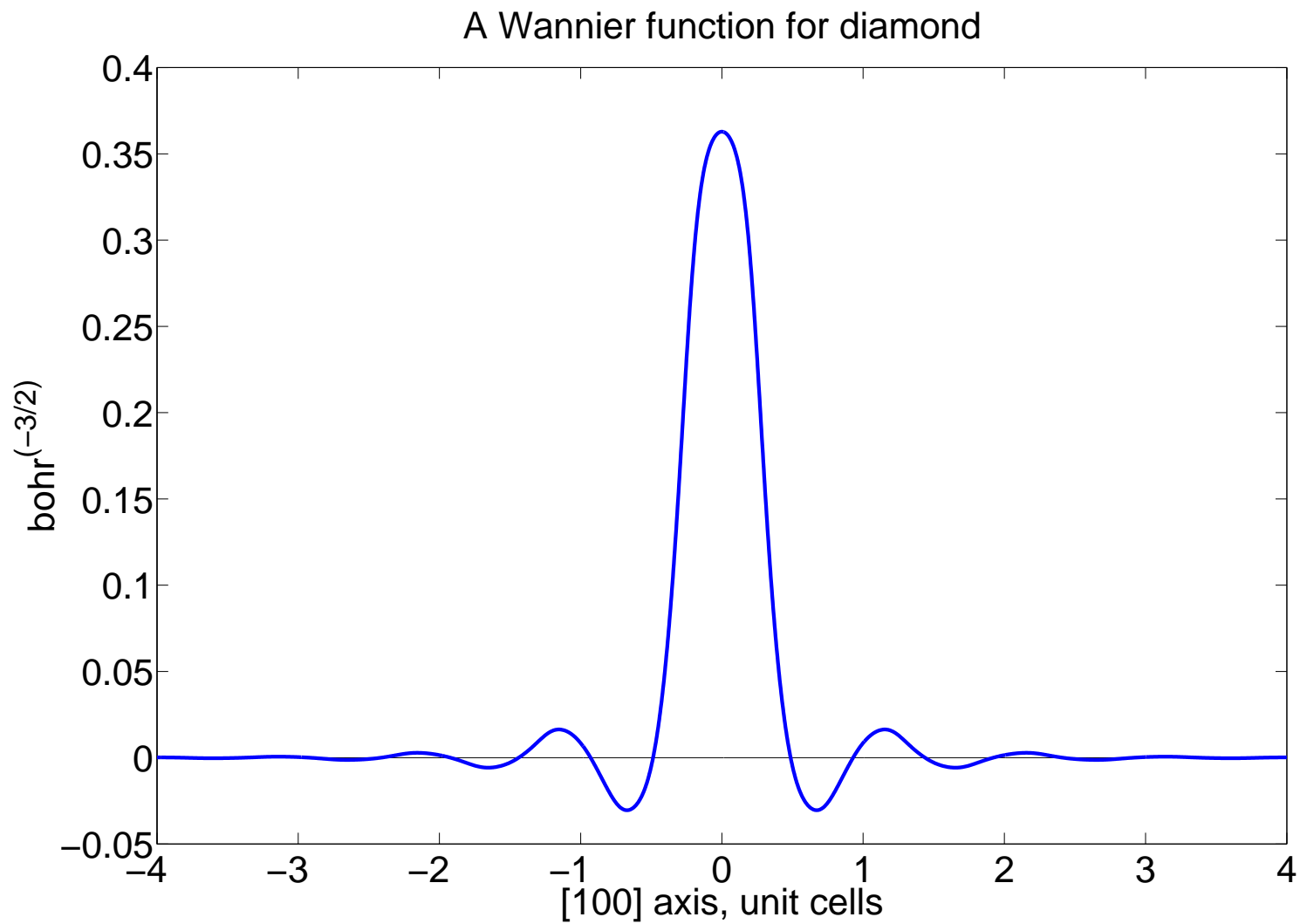
Conclusions and outlook

The DFP-LMP2 method, implemented in CRYSCOR code is evidently a **convenient and reliable tool** in studying weakly bound periodic systems.

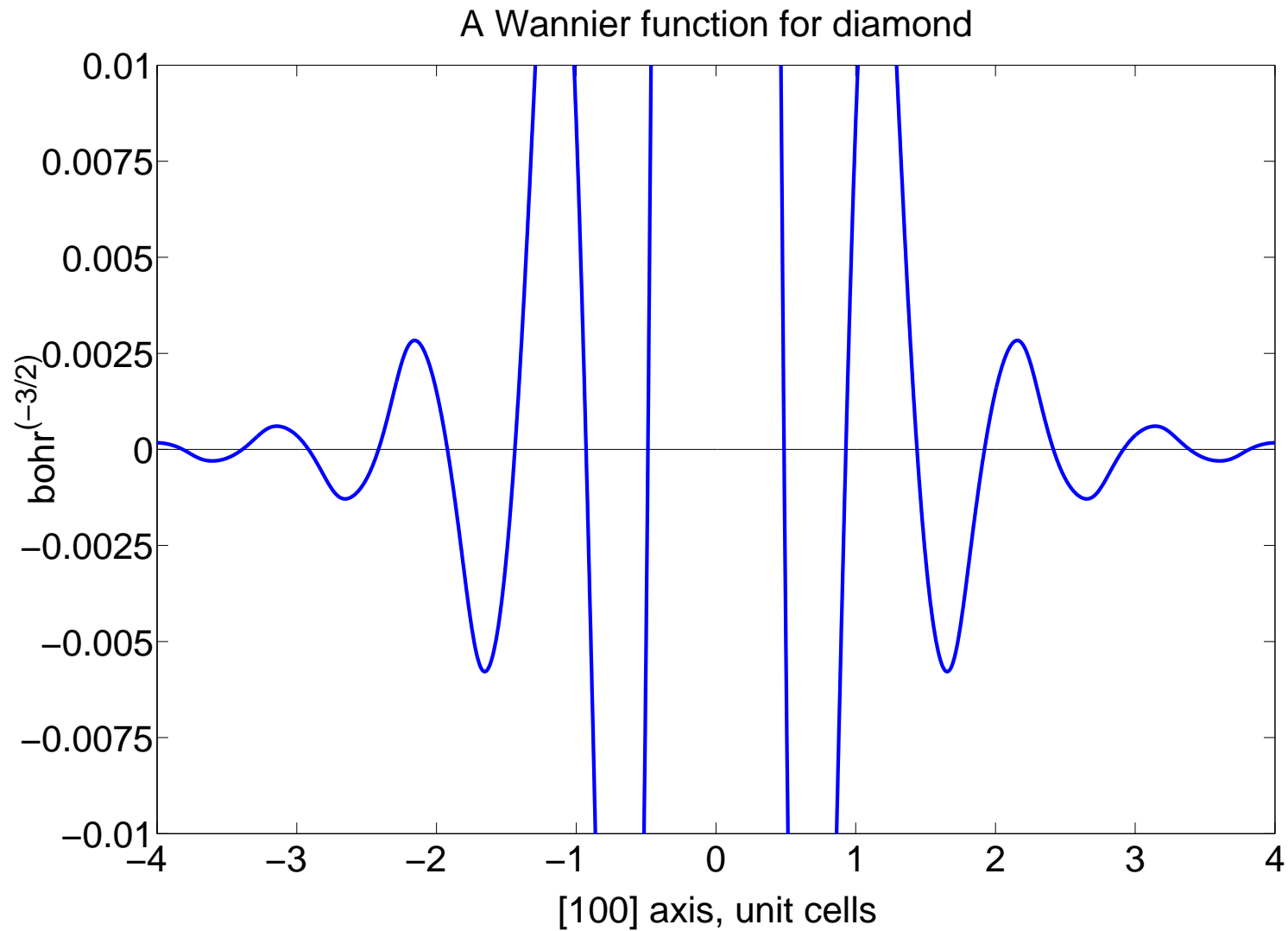
Problems:

- A need for larger basis sets \Rightarrow **dual basis sets, F12-technique** (work in progress).
- Poorly localized Wannier functions in conducting slabs \rightarrow Utilization of **finite-support Wannier functions**.

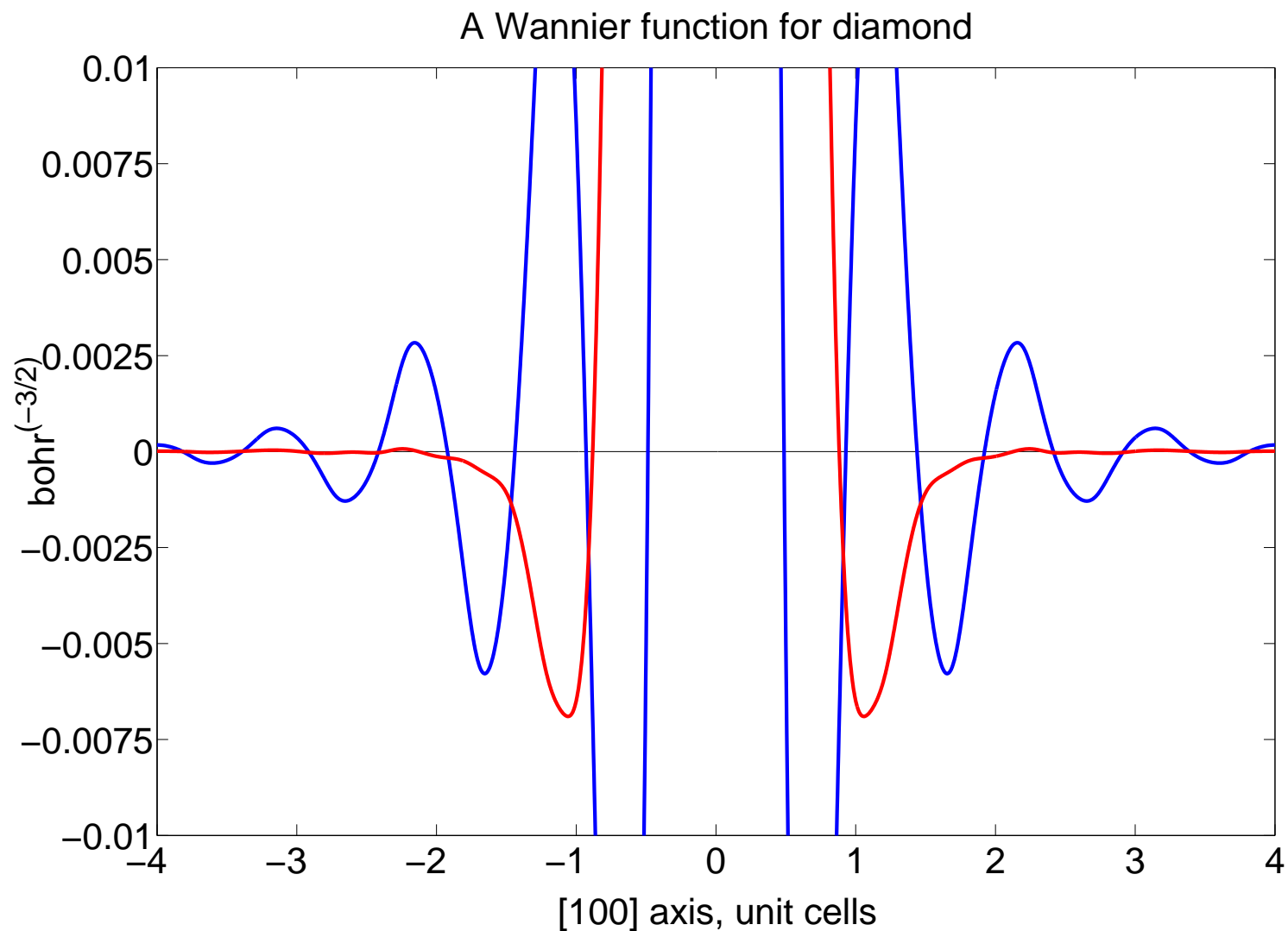
Localization of Wannier function



Localization of Wannier function



Localization of Wannier function



D. Usvyat and M. Schütz, TCA (2005)

Acknowledgments

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My collaborators:

- Regensburg theory group:
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- Cryscor Group in Torino:
Prof. Cesare Pisani
Dr. Lorenzo Maschio
Dr. Silvia Casassa Migen Halo

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